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## Determination of Ephedrines in Urine by HPLC

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### Abstract

An improved high-performance liquid chromatographic method with ultraviolet detection, for the simultaneous quantification of ephedrines (norephedrine, norpseudoephedrine, ephedrine, pseudoephedrine, methylephedrine and ethylephedrine) in urine is described. The six substances were separated on reversed phase column with phosphate buffer-TEA (pH 5.5) as mobile phase. The linearity and reproducibility was very satisfactory for levels usually found in urine (1-30 mg/l).

### Introduction

Ephedrine (EPH)-pseudoephedrine (PEPH), and norephedrine (NEPH) norpseudoephedrine (NPEP) are pairs of diastereoisomeric (1) compounds which are included in the doping list of pharmacological forbidden substances indicated by the Medical Commission of IOC (2). As they are widely available in asthma, ophthalmic, cold and allergy products (3) and as they are found in more than 100 pharmaceutical formulations, the mentioned Commission has set up concentrations above which they are considered positive. For this reason both, the determination of such pairs of diastereoisomers as well as their quantification are necessary.

The most often used technique for such determination and quantification is by gas chromatography coupled to mass spectrometry (4). In this case a previous selective derivatization (5) is needed (O-TMS,N-TFA), but not always the reproducibility obtained is enough because more than one derivative for the same compound is achieved. In this paper attempts have been made to develop a simple, rapid, selective and accurate HPLC method, where a previous derivatization process isn't necessary. It has been determined the two pairs of diastereoisomers and also methylephedrine (MEPH) and ethylephedrine (ETEP) are included (Fig.1).

Studies with positive urines proceeding from athletes have been carried out and very good results have been obtained.

## **Experimental**

### *Reagents*

NEPH, NPEP, EPH, PEPH, MEPH, ETEP were obtained from Sigma (St. Louis, Mo, USA). Diethylether was purchased from Carlo Erba (Milan, Italy). Phenylpropylamine (IS),  $\text{KH}_2\text{PO}_4$ ,  $\text{H}_3\text{PO}_4$  and triethylamine (TEA) were from E. Merck (Darmstadt, F.R.G.). Water was double-distilled, deionized and purified by Milli-Q system (Millipore, Milford, MA, USA). All other reagents and solvents were analytical-reagent grade.

### *Standard solutions and calibration standards*

Stock solutions of EPH, PEPH, NEPH, NPEP, MEPH, ETEP and IS were prepared in mobile phase described below at a concentration of 1000 mg/l. These stock solutions were then diluted further to yield appropriate working solutions for the preparation of the calibration standards. The solutions were sealed and refrigerated at 4°C until use.

### *Preparation of mobile phase*

The pH 5.5 buffer solution was adjusted by the appropriate addition of 200 mM phosphoric acid - 150 mM TEA to 200 mM potassium dihydrogen phosphate - 150 mM TEA. Before analysis, this mobile phase was filtered through 0.22  $\mu\text{m}$  filter and pumped through the column for 30 min. At the end of each chromatographic session, the column was washed for 15 min. with deionized water, and then with methanol.

### *Analytical procedure*

To 5 ml of urine in a 15 ml glass tube were added 25  $\mu\text{l}$  of IS (1000 mg/l), 100  $\mu\text{l}$  of 10 M NaOH and 2 ml of diethylether, then the urine was saturated with 3 g of Sodium sulphate and shook for 20 min. Tubes were centrifuged at 1200 g for 5 min and the organic layer was taken and evaporated to dryness. The extract was dissolved in 100  $\mu\text{l}$  of the mobile phase and 20  $\mu\text{l}$  of the solution was injected into the liquid chromatograph.

### *Chromatographic conditions*

The chromatographic system consisted of a Hewlett-Packard (Palo Alto, CA, USA) HP 1090 series, a liquid chromatograph equipped with an autosampler/autoinjector and an HP1040A diode-array UV detector. Chromatography was performed at 40° C on a (Hewlett-Packard) reversed-phase analytical column (Lichrospher 60 RP select B, 5  $\mu\text{m}$ , 125 mm X 4 mm I.D.). The mobile phase was 200 mM phosphate buffer (pH = 5.5) with

150 mM triethylamine at a flow-rate of 1.3 ml/min. The injection volume was 20  $\mu$ l and the column effluent was monitored at 215 nm (4 nm band width), where the ephedrines exhibit a maximum absorption. For data evaluation a HP 79994A Chemstation was used, which consisted of an HP 900 Series 300 computer, a 10 Mbytes Winchester disc drive and a Thinkjet printer.

## Results

Representative chromatograms for urine analyses are shown in Fig.2. The peaks of interest were well separated from potential interferences. Preliminary experiments led us to the conclusion that the separation depends significantly on the column used. The best results were obtained when a Select B column was used.

The composition of the mobile phase was selected in such a way that all the ephedrines were resolved in the shortest analysis time possible. As it is shown in Fig. 1, ephedrines have hydroxylated carbon at position B. For this reason they can interact with the free silanol groups. Using TEA in the mobile phase those interactions decreased because such silanol groups are masked (4). As long as the concentration of TEA of the mobile phase is increased, the  $K'$  values is decreased (Fig.3).

The pH of the mobile phase also influences on the determination of the six ephedrines (Fig.4). The  $K'$  values increased with increasing pH of the mobile phase, but a better resolution was achieved, above all with the pair of compounds pseudoephedrine and methylephedrine where the resolution was total at pH 5.5 ( $R_s=1.25$ ).

Precision and accuracy were carried out from spiked urine samples at concentration of 2, 8, 24 mg/l. The samples were extracted and subjected to HPLC. Each concentration was calculated on bases of the peak-height ratios against the IS. The results are listed in Table I. The linearity was evaluated over the range of concentrations from 1 to 30 mg/l using duplicate spike samples at levels of 1, 5, 10, 15 and 30 mg/l. The calibration curves obtained and the equations of the mean plots are shown in Fig. 5 and Table II, respectively.

The limit of detection was defined as the lowest concentration of each ephedrine resulting in a single-to-noise ratio of three. For NEPH, NPEP and EPH the limit was 0.2 mg/l and for PEPH, MEPH and ETEP their limit was 0.5 mg/l.

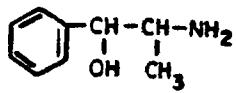
Other pharmacological substances were examined in order to establish possible interferences. Substances tested are shown in Table III. None of these interfered the ephedrines determination.

## Conclusion

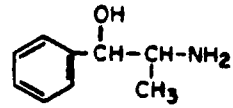
In conclusion, the elution and separation of ephedrins were clearly affected by the column used and the concentration of TEA as well as pH of the mobile phase. As it was not necessary the use of a modifier, endogenous compounds were eluted in very long times resulting very clear chromatograms where interferences of such endogenous compounds weren't observed. This method has been used in urines proceeding from athletes, obtaining very satisfactory results.

## References

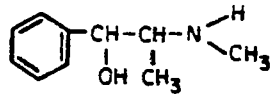
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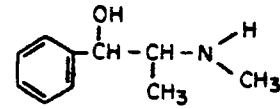
Norephedrine



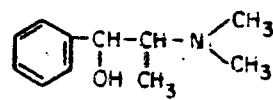
Norpseudoephedrine



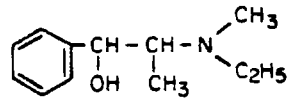
Ephedrine



Pseudoephedrine



N-Methylephedrine



Ethylephedrine

Fig. 1

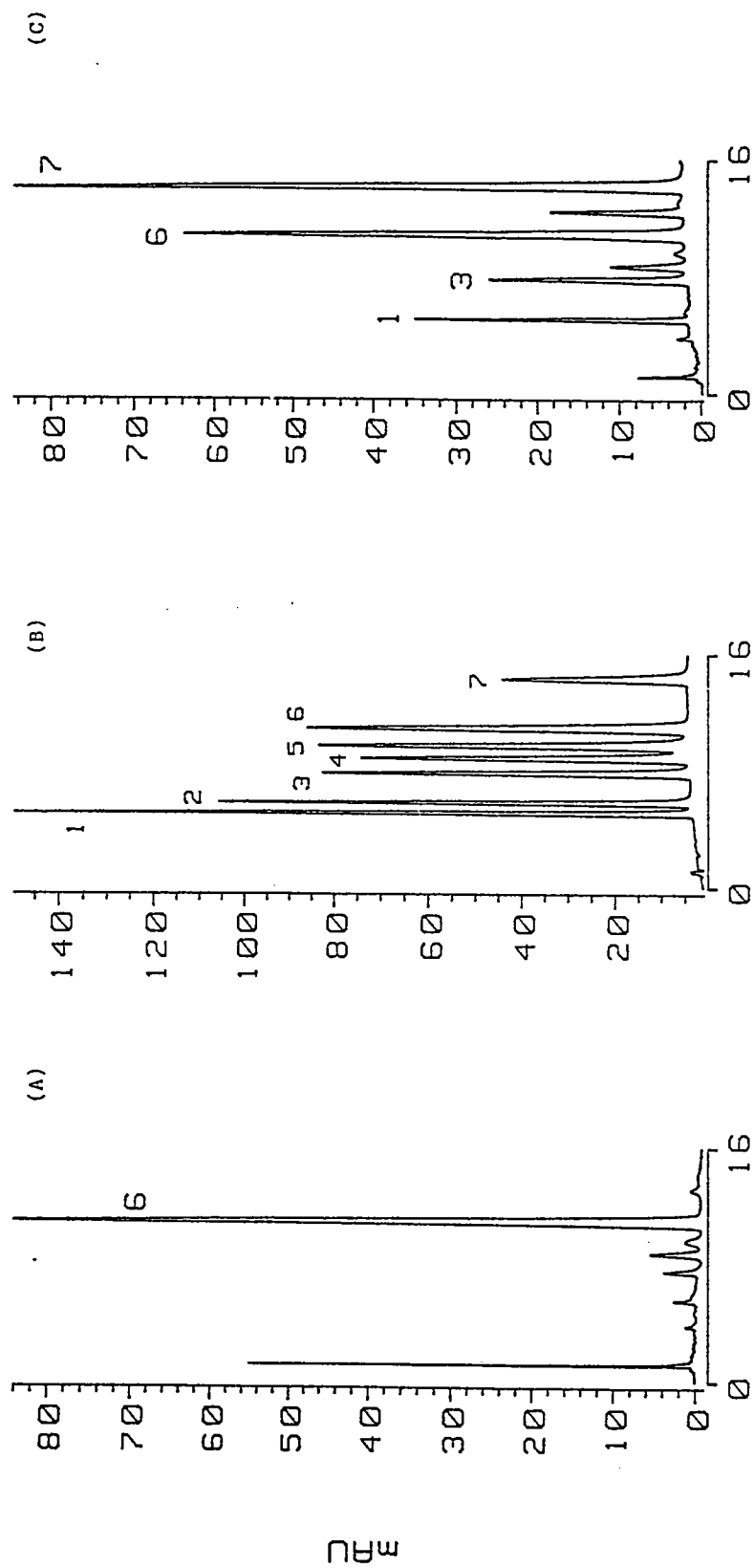


Fig. 2. Chromatograms obtained after analysis of (A) blank urine to which internal standard (peak 6) was added, (B) urine standard spiked with NEPH (peak 1), NPEP (peak 2), EPH (peak 3), PEPH (peak 4), MEPH (peak 5), ETEP (peak 7) and IS (peak 6), and (C) urine sample from an athlete who took ETEP.

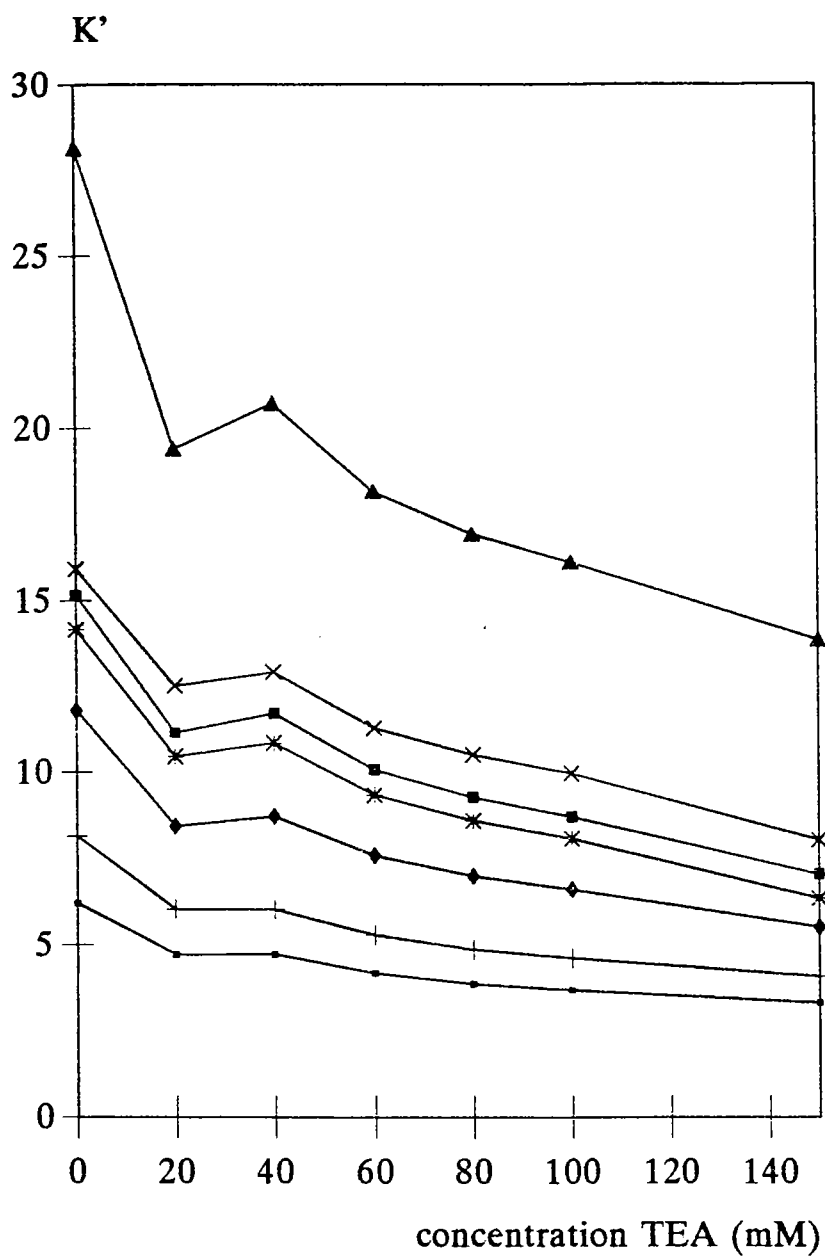


Fig. 3. Effect of the concentration of TEA in the column on  $K'$  values of ephedrines. Mobile phase, 200 mM phosphate (pH 4); column, lichrospher 60 RP Select B, 5  $\mu$ m (125 mm x 4 mm I.D.)

→ NEPH + NPEPH \* PEPH ■ MEPH × IS ◆ EPH ▲ ETEPH



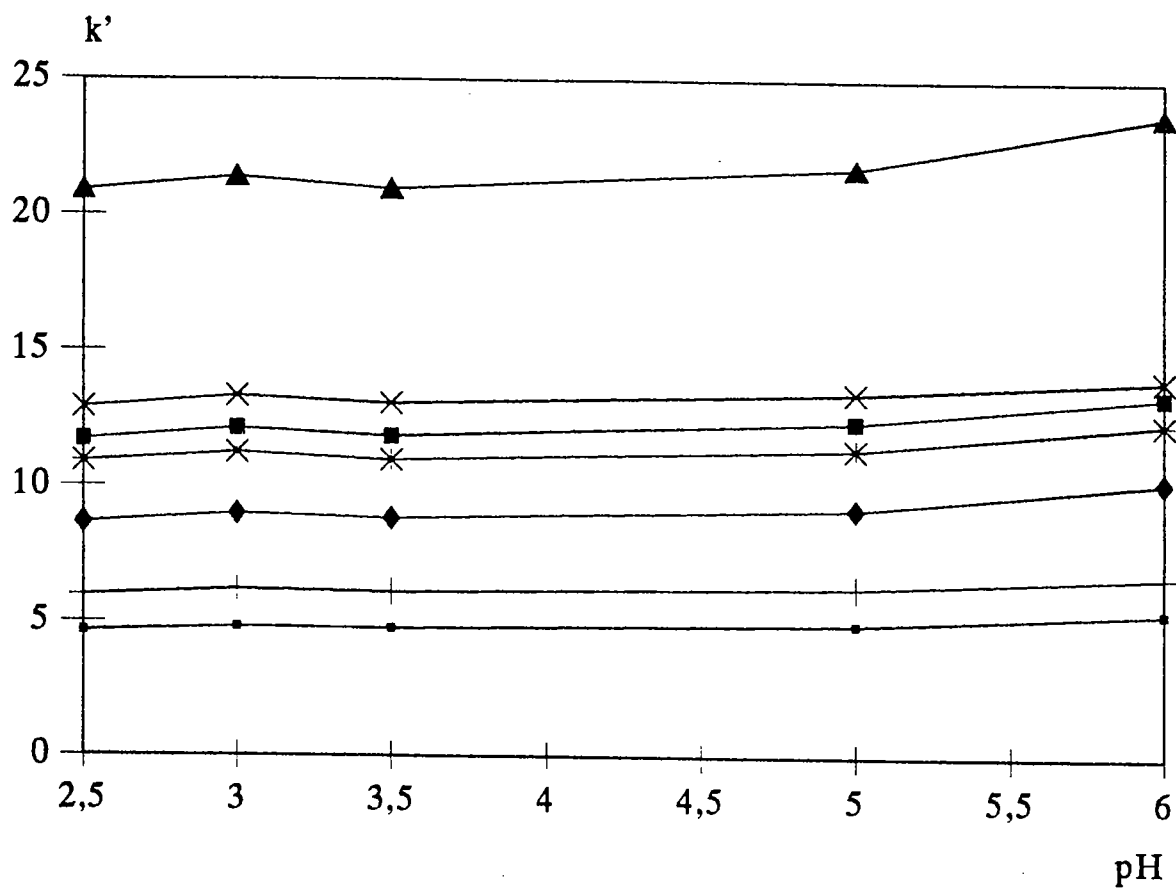


Fig. 4. Effect of pH on  $K'$  values. Concentration of TEA of the mobile phase, 150 mM; other conditions and compounds as specified in Fig. 3.

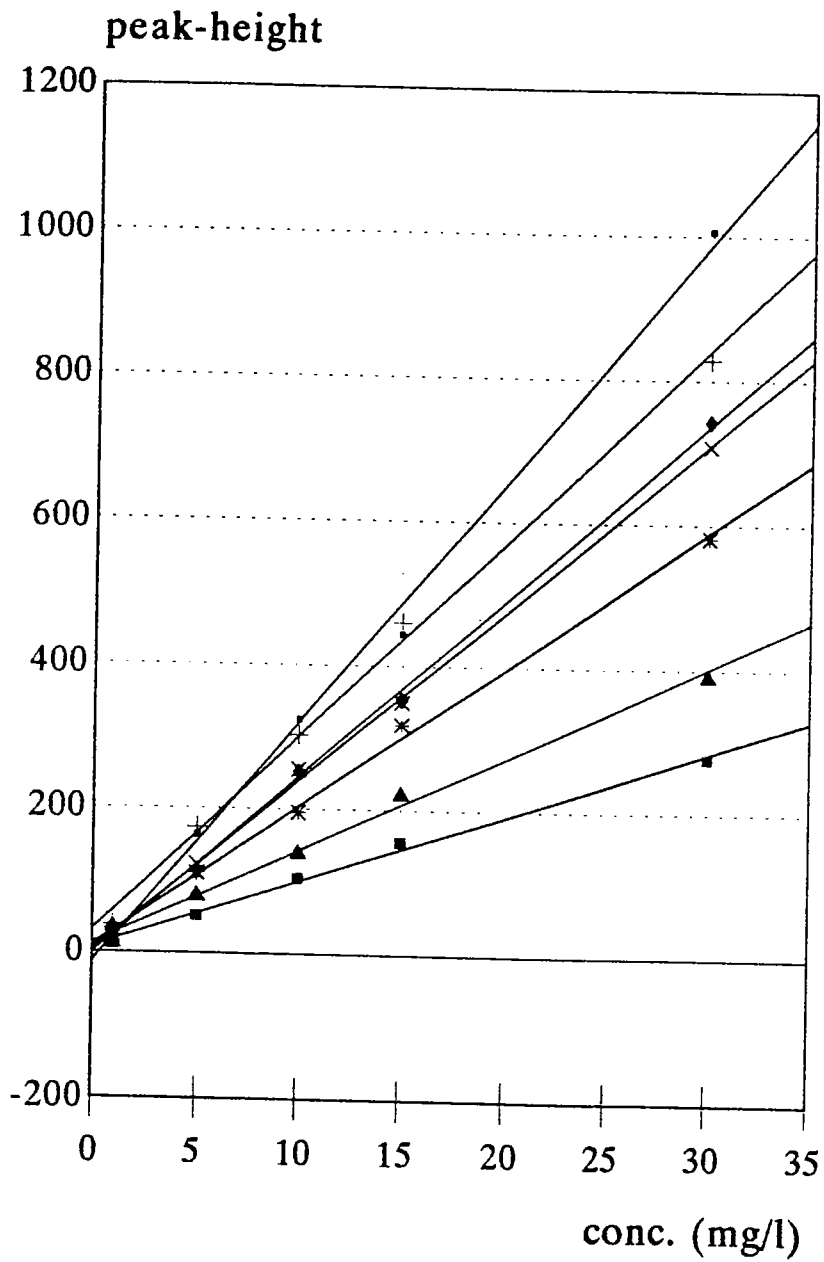


Fig. 5. Linear calibration curves for the ephedrines specified in

ANALYTICAL ACCURACY AND REPRODUCIBILITY OF THE ANALYSIS OF EPHEDRINES IN HUMAN URINE

Concentration Added ( $\mu\text{g/ml}$ )	NEPH		NPEP		EPH		PEPH		MEPH		EPEP	
	Mean $\pm$ S.D. ( $\mu\text{g/ml}$ )	C.V. (%)	Mean $\pm$ S.D. ( $\mu\text{g/ml}$ )	C.V. (%)	Mean $\pm$ S.D. ( $\mu\text{g/ml}$ )	C.V. (%)	Mean $\pm$ S.D. ( $\mu\text{g/ml}$ )	C.V. (%)	Mean $\pm$ S.D. ( $\mu\text{g/ml}$ )	C.V. (%)	Mean $\pm$ S.D. ( $\mu\text{g/ml}$ )	C.V. (%)
2	1.98 <sup>+</sup> -0.07	3.5	2.05 <sup>+</sup> -0.03	1.3	1.96 <sup>+</sup> -0.09	2.2	1.92 <sup>+</sup> -0.02	1.0	1.97 <sup>+</sup> -0.05	2.69	1.87 <sup>+</sup> -0.12	3.1
8	8.04 <sup>+</sup> -0.05	5.3	8.22 <sup>+</sup> -0.06	0.7	8.27 <sup>+</sup> -0.04	0.4	8.07 <sup>+</sup> -0.06	0.7	7.95 <sup>+</sup> -0.04	0.56	7.23 <sup>+</sup> -0.21	4.4
24	25.93 <sup>+</sup> -0.26	0.9	24.34 <sup>+</sup> -0.28	1.1	26.00 <sup>+</sup> -1.17	4.5	25.59 <sup>+</sup> -0.23	0.9	22.47 <sup>+</sup> -0.28	1.26	25.14 <sup>+</sup> -1.49	6.4

Table I.

## EQUATIONS OF LINEAR CALIBRATION CURVES

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NEPH	$y=33.38x-13.63$	$rr=0.9978$
NPEP	$y=27.02x+30.64$	$rr=0.9983$
EPH	$y=24.58x-0.92$	$rr=0.9991$
PEPH	$y=19.27x+10.82$	$rr=0.9987$
MEPH	$y=9.12x+8.78$	$rr=0.9975$
IS	$y=23.55x+3.66$	$rr=0.9993$
ETEP	$y=12.93x+13.08$	$rr=0.9971$

Tabla II.

## **DRUGS TESTED FOR POSSIBLE INTERFERENCE**

HEPTAMINOL  
AMPHETAMINE  
METHYLAMPHETAMINE  
FENCAMFAMINE  
DIMETHYLAMPHETAMINE  
NICOTINE  
PHENDIMETRAZINE  
METHOXAMINE  
LEPTAZOL  
NIKETAMINE  
PETIDINE  
CAFFEINE  
LIDOCAINE  
PROCAINE  
CODEINE  
STRYCHNINE  
ETAMIVAN  
PHENMETRAZINE  
METHYLPHENIDATE  
PROLINTANE  
PIPRADOL  
AMFEPRAMONE  
CHLORPHENTERMINE  
COCAINE

Table III.